?-Aromaticity-Induced Stabilization of Heterometallic Supertetrahedral Clusters [Zn6Ge16]4? and [Cd6Ge16]4?

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In this work, the largest heterometallic supertetrahedral clusters, [Zn6Ge16]4? and [Cd6Ge16]4?, were directly self-assembled through highly-charged [Ge4]4? units and transition metal cations, in which 3-center?2-electron ? bonding in Ge2Zn or Ge2Cd triangles plays a vital role in the stabilization of the whole structure. The cluster structures have an open framework with a large central cavity of diameter 4.6 Å for Zn and 5.0 Å for Cd, respectively. Time-dependent HRESI-MS spectra show that the larger clusters grow from smaller components with a single [Ge4]4? and ZnMes2 units. Calculations performed at the DFT level indicate a very large HOMO?LUMO energy gap in [M6Ge16]4? (2.22 eV), suggesting high kinetic stability that may offer opportunities in materials science. These observations offer a new strategy for the assembly of heterometallic clusters with high symmetry. © 2020 Wiley-VCH GmbH

planar tetracoordinate Zn/Cd

supertetrahedron

Td symmetry

Zintl clusters

?-aromaticity

Cadmium compounds

Stabilization

Transition metals

- Zinc compounds
- Cluster structure

Heterometallic clusters

- Heterometallics
- Kinetic stability
- Open frameworks
- Sigma aromaticity
- Time dependent
- Transition metal cations
- Germanium compounds