

Incorporation of Nanostructured ReO_3 in Silica Matrix and Their Activity Toward Photodegradation of Blue Methylene

Diaz C.

Valenzuela M.L.

Cifuentes-Vaca O.

Segovia M.

Laguna-Bercero M.A.

Abstract: ReO_3 were prepared by thermal treatment of the macromolecular Chitosan·(ReCl_3) $_x$ and PSP-4-PVP·(ReCl_3) $_x$ precursors. The plasmon band in the visible region for the as obtained ReO_3 from their visible spectra was observed at λ_{max} of 640 nm. The nature of the polymeric precursor is acting as a solid state template and influences the size and morphology of the metal oxides. For the first time, the photocatalytic degradation of methylene blue using ReO_3 was measured founding a moderated and high activity for ReO_3 arise from Chitosan and PSP-4-PVP precursors respectively. The inclusion of ReO_3 into SiO_2 was performed using a combined solution of the Chitosan and PVP precursors by the sol-gel method. Subsequent pyrolysis of the solid precursors Chitosan·(ReCl_3) $_x$ (SiO_2) $_y$ and PSP-4-PVP·(ReCl_3) $_x$ (SiO_2) $_y$ give rise to the nanocomposites $\text{ReO}_3/\text{SiO}_2$. The as obtained ReO_3 nanoparticles inside SiO_2 are small as 1 nm. The ReO_3 nanoparticles are distributed uniformly inside the matrix of SiO_2 , leading to stable semi porous materials suitable for high temperature catalytic application. The composites $\text{ReO}_3/\text{SiO}_2$ exhibit a moderate photocatalytic activity toward the degradation of methylene blue and similar to that of ReO_3 . **Graphic Abstract:** [Figure not available: see fulltext.]. © 2019, Springer Science+Business Media, LLC, part of Springer Nature.

Blue methylene

Nanostructured ReO_3

Photodegradation

Aromatic compounds

Chitosan

Nanocomposites

Nanoparticles

Photocatalytic activity

Photodegradation

Porous materials

Silica

SiO₂ nanoparticles

Sols

Blue methylene

Catalytic applications

Combined solution

High temperature

Nano-structured

Photo catalytic degradation

Polymeric precursors

Solid precursors

Rhenium compounds