
Title

Adsorption of a Ni single atom catalyst on a pristine nanographene (coronene) and H-H activation. A theoretical study of Ni-SAC

Abstract

A computational study of a Ni atom in a coronene (Cor) molecule, as a model of a single-atom catalyst (SAC), was carried out using a DFT approach. The adsorption energy results of Ni on Cor (2.31–1.84 eV) are similar to those reported for extended graphene. Ni-Cor adsorption is preferred at edge sites and electron transfer from Ni to Cor occurs. H₂ adsorbs on Ni-Cor with 1.03–1.24 eV. Correlations were obtained between the H₂ adsorption energies and some properties of Ni-Cor. The results suggest that Ni-Cor could be used for hydrogen storage and also for catalytic hydrogenation of hydrocarbons. © 2024

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References

Su C., Loh K.P., Carbocatalysts: graphene oxide and its derivatives, *Acc. Chem. Res.*, 46, pp. 2275-2285, (2013); Ren W., Cheng H.M., The global growth of graphene, *Nat. Nanotechnol.*, 9, pp. 726-730, (2014); Julkapli N.M., Bagheri S., Graphene supported heterogeneous catalysts: an overview, *Int. J. Hydrogen Energy*, 40, pp. 948-979, (2015); Durgun E., Dag S., Bagci V.M.K., Gulseren O., Yildirim T., Ciraci S., Systematic study of adsorption of single atoms on a carbon nanotube, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 67, pp. 1-4, (2003); Yagi Y., Briere T.M., Sluiter M.H.F., Kumar V., Farajian A.A., Kawazoe Y.; Ishii A., Yamamoto M., Asano H., Fujiwara K., DFT calculation for adatom adsorption on graphene sheet as a prototype of carbon nano tube functionalization, *J. Phys. Conf. Ser.*, 100, (2008); Rigo V.A., Martins T.B., Silva A.J.R., Fazzio A., Miwa R.H.; Nakada K., Ishii A., pp. 3-20, (2011); Mashhadzadeh A.H., Vahedi A.M., Ardjmand M., Ahangari M.G., Investigation of heavy metal atoms adsorption onto graphene and graphdiyne surface: A density functional theory study, *Superlattice. Microst.*, (2016); Manade M., Vines F., Illas F., Transition metal adatoms on graphene: A systematic density functional study, *Carbon N Y*, 95, pp. 525-534, (2015); Hernandez-Vazquez E.E., Munoz F., Lopez-Moreno S., Moran-Lopez J.L., First-principles study of Ni adatom migration on graphene with vacancies, *RSC Adv.*, 9, pp. 18823-18834, (2019); Nishchakova A.D., Bulusheva L.G., Bulushev D.A., Supported Ni single-atom catalysts: synthesis, structure, and applications in thermocatalytic reactions, *Catalysts*, 13, (2023); Hossain M.D., Huang Y., Yu T.H., Goddard W.A., Luo Z., Reaction mechanism and kinetics for CO₂ reduction on nickel single atom catalysts from quantum mechanics, *Nat. Commun.*, 11, (2020); Jiang K., Siahrostami S., Zheng T., Hu Y., Hwang S., Stavitski E., Et al., Isolated Ni single atoms in graphene nanosheets for high-performance CO₂ reduction, *Energ. Environ. Sci.*, 11, pp. 893-903, (2018); Xu Y., Zhang W., Li Y., Lu P., Wu Z.S., A general bimetal-ion adsorption strategy to

prepare nickel single atom catalysts anchored on graphene for efficient oxygen evolution reaction, *J. Energy Chem.*, 43, pp. 52-57, (2020); Song X., Li N., Zhang H., Wang L., Yan Y., Wang H., Et al., Applications graphene-supported single nickel atom catalyst for highly selective and efficient hydrogen peroxide production, *Energy, Environ., Catal.*, (2020); Gao Z., Li L., Huang H., Xu S., Yan G., Zhao M., Et al., Adsorption characteristics of acid gases (NO, NO₂, SO₂, and SO₃) on different single-atom nickel adsorbent: a first-principles study, *Appl. Surf. Sci.*, 527, (2020); Qiu H.J., Ito Y., Cong W., Tan Y., Liu P., Hirata A., Et al., Nanoporous graphene with single-atom nickel dopants: an efficient and stable catalyst for electrochemical hydrogen production, *Angew. Chem. – Int. Ed.*, 54, pp. 14031-14035, (2015); Fan L., Liu P.F., Yan X., Gu L., Yang Z.Z., Yang H.G., Et al., Atomically isolated nickel species anchored on graphitized carbon for efficient hydrogen evolution electrocatalysis, *Nat. Commun.*, 7, (2016); Wang M., Wang Z., Single Ni atom incorporated with pyridinic nitrogen graphene as an efficient catalyst for CO oxidation: first-principles investigation, *RSC Adv.*, 7, pp. 48819-48824, (2017); Sanchez M., Ruetten F., Calculations of adsorption, coadsorption, diffusion, and reaction barriers of H atoms in the H₂ formation on a positively charged coronene, *Chem. Phys. Lett.*, 738, (2020); Chen Y., Habibullah X.G., Jin C., Wang Y., Yan Y., Et al., Hydrogen storage properties of economical graphene materials modified by non-precious metal nickel and low-content palladium, *Inorganics (Basel)*, 11, (2023); Cui H., Yan C., Jia P., Cao W., Adsorption and sensing behaviors of SF₆ decomposed species on Ni-doped C₃N monolayer: a first-principles study, *Appl. Surf. Sci.*, 512, (2020); Kumar S., Bukkigar S.D., Singh S., Pratibha S.V., Reddy K.R., Et al., Electrochemical sensors and biosensors based on graphene functionalized with metal oxide nanostructures for healthcare applications, *ChemistrySelect*, 4, pp. 5322-5337, (2019); Neese F., Software update: The ORCA program system—Version 5.0, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 12, (2022); Ernzerhof M., Scuseria G.E., Assessment of the Perdew–Burke–Ernzerhof exchange–correlation functional, *J. Chem. Phys.*, 110, pp.

5029-5036, (1999); Steinmetz M., Grimme S., Benchmark study of the performance of density functional theory for bond activations with (Ni, Pd)-based transition-metal catalysts, *ChemistryOpen*, 2, pp. 115-124, (2013); Harrison J.F., Electronic structure of diatomic molecules composed of a first-row transition metal and main-group element (H-F), *Chem. Rev.*, 100, pp. 679-716, (2000); Rao P.S., Reddy R.R., Potukuchi D.M., Estimation of dissociation energy of NiC molecule, *J. Quant. Spectrosc. Radiat. Transf.*, 98, pp. 81-84, (2006); Oluwadare O.J., Oyewumi K.J., Energy spectra and the expectation values of diatomic molecules confined by the shifted Deng-Fan potential, *Eur. Phys. J. plus*, 133, (2018); Meyer F., Khan F.A., Armentrout P.B., (1995); Jaeger T.D., Van Heijnsbergen D., Klippenstein S.J., Von Helden G., Meijer G., Duncan M.A., Vibrational spectroscopy and density functional theory of transition-metal ion - Benzene and dibenzene complexes in the gas phase, *J. Am. Chem. Soc.*, 126, pp. 10981-10991, (2004); Klippenstein S.J., Yang C.-N., Density functional theory predictions for the binding of transition metal cations to pi systems: from acetylene to coronene and tribenzocyclyne, *Int. J. Mass Spectrom.*, 201, pp. 253-267, (2000); Dunbar R.C., Binding of transition-metal ions to curved π surfaces: Corannulene and coronene, *J. Phys. Chem. A*, 106, pp. 9809-9819, (2002); Glendening E.D., Landis C.R., Weinhold F., NBO 6.0: natural bond orbital analysis program, *J. Comput. Chem.*, 34, pp. 1429-1437, (2013); Valencia H., Gil A., Frapper G., Trends in the adsorption of 3d transition metal atoms onto graphene and nanotube surfaces: A DFT study and molecular orbital analysis, *J. Phys. Chem. C*, 114, pp. 14141-14153, (2010); Eelbo T., Waaniowska M., Thakur P., Gyamfi M., Sachs B., Wehling T.O., Et al., Adatoms and clusters of 3d transition metals on graphene: electronic and magnetic configurations, *Phys. Rev. Lett.*, 110, pp. 3-7, (2013); Wehling T.O., Lichtenstein A.I., Katsnelson M.I.; Lima M.P.; Longo R.C., Carrete J., Ferrer J., Gallego L.J., Structural, magnetic, and electronic properties of N_{in} and F_{en} nanostructures ($n=1-4$) adsorbed on zigzag graphene nanoribbons, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 81, (2010); Liu X., Wang C.Z., Yao Y.X.,

Lu W.C., Hupalo M., Tringides M.C., Et al.; Duffy D.M., Blackman J.A., Magnetism of 3d transition-metal adatoms and dimers on graphite, *Phys. Rev. B*, 58, pp. 7443-7449, (1998); Sigal A., Rojas M.I., Leiva E.P.M., Interferents for hydrogen storage on a graphene sheet decorated with nickel: A DFT study, *Int. J. Hydrogen Energy*, 36, pp. 3537-3546, (2011); Manade M., Vines F., Gil A., Illas F., On the H₂ interactions with transition metal adatoms supported on graphene: a systematic density functional study, *PCCP*, 20, pp. 3819-3830, (2018); Aal S.A., Soliman A.S.S.K.A., High capacity hydrogen storage in ni decorated carbon nanocone: a first-principles study, *J. Quantum Inf. Sci.*, 5, pp. 134-149, (2015); Ruetter F., Sanchez M., Castellano O., Soscun H., Methodologies to analyze surface bonding properties using parametric and density functional methods, *Int. J. Quantum Chem*, 110, pp. 743-754, (2010); Kubas G.J., Molecular Hydrogen Complexes: Coordination of a σ Bond to Transition Metals, *Acc. Chem. Res.*, 21, pp. 120-128, (1988); Yang X., Cheng J., Fang B., Xuan X., Liu N., Yang X., Et al., Single Ni atoms with higher positive charges induced by hydroxyls for electrocatalytic CO₂ reduction, *Nanoscale*, 12, pp. 18437-18445, (2020); Serny K.A., Ruetter F., Camacho J.M., Vega C., Inferential Statistics using Partial Least Square Regression (PLSR) for Chemical Processes. Case Study : A Fixed Bed Catalytic Reactor, *J. Stat. Math. Eng.*, 6, pp. 14-23, (2020); Nisha S., Senthil K.A., π -Self-Assembly of a Coronene on Carbon Nanomaterial-Modified Electrode and Its Symmetrical Redox and H₂O₂ Electrocatalytic Reduction Functionalities, *ACS Omega*, 5, pp. 11817-11828, (2020)

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