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## Title

***Assessing the electronic properties of bimetallic complexes with N-M-N-M cycle (M = Ag, Au and Cu)***

## Abstract

This work is a theoretical study of the electronic properties of the reported bimetallic complexes  $M_2(Ms(CH_2SCN)_3)_2$  (where  $M = Cu^+, Ag^+, Au^+$ ) coordinate with  $Ms(CH_2SCN)_3$  ligand acting simultaneously as a bidentate chelate and bridge. The electronic properties were studied using reactivity indexes, Fukui function, NICS indexes showing that the metal is susceptible to a nucleophilic attack, whereas the nitrogen and sulfur ligand are susceptible to an electrophilic attack, and that the Phenyl ring aromaticity does not change with the metallic size. The absorption properties were studied via TD-DFT methods incorporating the relativistic scalar and spin-orbit correction. The most intense transitions are mainly localized in the visible region of the spectrum with a Metal to Ligand Charge Transfer (MLCT) character. The metal/ligand interactions were studied via the Morokuma-Ziegler decomposition scheme and Natural Orbital of Chemical Valence (NOCV). The three complexes might act as photocatalyst due to their visible light absorption observed transitions.

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## References

Dariusz M.S., Szczepanik W., The electron density of delocalized bonds (EDDBs) as a measure of local and global aromaticity, *Phys. Chem. Chem. Phys.*, (2021); Ferran Feixas M.S., Matito E., Poater J., Understanding conjugation and hyperconjugation from electronic delocalization measures, *J. Phys. Chem. A*, 115, pp. 13104-13113, (2011); Igor P.W.P., Alabugin V., Gilmore K.M., Hyperconjugation, *Advanced Review*, 1, pp. 109-141, (2011); Li X., Kuznetsov A.E., Zhang H.-F., Boldyrev A.I., Wang A.-S., Observation of all-metal aromatic molecules, *Science*, 291, 2001, (1979); Biju Basumatary J.S., Jyoti Rai R.V., Reddy R., Evidence for a [17] p-electronic full-fledged non-innocent Gallium(III)-Corrole radical, *Chem. Eur. J.*, 23, pp. 17458-17462, (2017); Datta A., Pati S.K., Stable transition metal complexes of an all-metal antiaromatic molecule ( $\text{Al}_4\text{Li}_4$ ): role of complexations, *J. Am. Chem. Soc.*, 127, (2005); Shu Li Q., Jin Q., Aromaticity of planar  $\text{B}_5^-$  anion in the  $\text{MB}_5$  ( $M = \text{Li, Na, K, Rb, and Cs}$ ) and  $\text{MB}_5^+$  ( $M = \text{Be, Mg, Ca, and Sr}$ ) clusters, *J. Phys. Chem. A*, 127, (2004); Daisuke Yokogawa S.S., Sato H., Nakao Y., Localization or delocalization in the electronic structure of Creutz-Taube-type complexes in aqueous solution, *Inorg. Chem.*, 46, pp. 1966-1974, (2007); Marzena Marchaj P.S., Freza S., Why Are  $\text{SiX}_5^-$  and  $\text{GeX}_5^-$  ( $X = \text{F, Cl}$ ) Stable but Not  $\text{CF}_5^-$  and  $\text{CCl}_5^-$ ?, *J. Phys. Chem. A*, 116, pp. 1966-1973, (2012); Alvarado-Soto L., Ramirez-Tagle R., Arratia-Perez R., Spin-orbit effects on the aromaticity of the  $\text{Re}_3\text{Cl}_9$  and  $\text{Re}_3\text{Br}_9$  clusters, *Chem. Phys. Lett.*, 467, (2008); Chen Z., Wannere C.S., Corminboeuf C., Puchta R., von Rague Schleyer P., Nucleus-independent chemical shifts (NICS) as an aromaticity criterion, *Chem. Rev.*, 105, pp. 3842-3888, (2005); You F., Qiu R., Zhu J.,

---

Adaptive aromaticity in osmapentalene and osmapyridinium complexes with carbene ligands, *J. Phys. Org. Chem.*, 36, (2023); Akhilesh Krishnan D.C., Diaz-Andres A., Sudhakaran K.P., John A.T., Hariharan M., Deciphering the role of (anti)aromaticity in cofacial excimers of linear acenes, *J. Phys. Org. Chem.*, 36, (2023); Arias-Olivares D., Paez-Hernandez D., The spin-orbit effects on platinabenzene: a ring current and electron delocalization approach, *New J. Chem.*, 46, pp. 16708-16716, (2022); Alvarez C.M., Angeles Alvarez M., Esther Garcia M., Ramos A., Ruiz M.A., Graiff C., Tiripicchio A., Reactivity of the unsaturated hydride  $[\text{Mo}_2(\eta^5\text{-C}_5\text{H}_5)_2(\mu\text{-H})(\mu\text{-PCy}_2)(\text{CO})_2]$  toward 17- and 16-electron metal carbonyl fragments : Rational synthesis of electron-deficient heterometallic clusters, *Organometallics*, 26, pp. 1461-1472, (2007); Jimenez-Halla J.O.C., Matito E., Robles J., Sola M., Nucleus-independent chemical shift (NICS) profiles in a series of monocyclic planar inorganic compounds, *J. Organometallic Chem.*, 691, pp. 4359-4366, (2006); Gomes J.A.N.F., Mallion R.B., Aromaticity and ring currents, *Chem. Rev.*, 101, pp. 1349-1384, (2001); Gabriel Merino G.S., Heine T., The induced magnetic field in cyclic molecules, *Chem Eur J*, 10, pp. 4367-4371, (2004); Cina Foroutan-Nejad P.-R.-R., Badri Z., Shahbazian S., The Laplacian of electron density versus NICS<sub>zz</sub> Scan: measuring magnetic aromaticity among molecules with different atom types, *J. Phys. Chem. A*, 115, pp. 12708-12714, (2011); Tshipis A.C., Efficiency of the NICS<sub>zz</sub>-scan curves to probe the antiaromaticity of organic and inorganic rings/cages, *Phys. Chem. Chem. Phys.*, 2009, 11, pp. 8244-8261, (2009); Paez-Hernandez D., Arratia-Perez R., Aromaticity, optical properties and zero field splitting of homo- and hetero-bimetallic  $(\text{C}_8\text{H}_8)_2\text{M}(\mu_2\text{-}\eta^8\text{-C}_8\text{H}_8)_2\text{M}(\text{C}_8\text{H}_8)$  where M = Ti, Zr, Th complexes, *Journal of Physical Chemistry A*, 116, pp. 7584-7592, (2012); Eduardo Schott R.-A.-P., Zarate X., Relativistic scalar and spin-orbit density functional

---

calculations of the electronic structure, NICS index and ELF function of the  $[\text{Re}_2(\text{CO})_8(\text{l-BiPh})_2]$  and  $[\text{Re}_2(\text{CO})_8(\text{l-BiPh})_2]_2$  clusters, *Polyhedron*, 30, pp. 846-850, (2011); Adams R.D., Pearl W.C., Jr., Rhenium–Bismuth carbonyl cluster compounds, *Inorg. Chem.*, 48, pp. 9519-9525, (2009); Hector Martinez-Garcia D.M., Morales D., Perez J., Puerto M., 1,3,5-Tris(thiocyanatomethyl)mesitylene as a ligand. Pseudooctahedral molybdenum, manganese, and rhenium carbonyl complexes and copper and silver dimers. Copper-catalyzed carbene- and nitrene-transfer reactions, *Inorg. Chem.*, 49, pp. 6974-6985, (2010); Baerends S.J., Ziegler T., Autschbach J., Bashford D., Berces A., Bickelhaupt F.M., Bo C., Boerrigter P.M., Cavallo L., Chong D.P., Deng L., Dickson R.M., Ellis D.E., (2017); van Lenthe E., Baerends E.J., Snijders J.G., Relativistic regular two-component Hamiltonians, *J. Chem. Phys.*, 99, pp. 4597-4610, (1993); Snijder P.V.J.G., Baerends E.J., Roothaan-Hartree-Fock-Slater atomic wave functions: Single-zeta, double-zeta, and extended Slater-type basis sets for 87Fr-103Lr, *At. Data Nucl. Data Tables*, 26, pp. 483-509, (1982); Verluis T.Z.L., The determination of molecular structures by density functional theory. the evaluation of analytical energy gradients by numerical Integration, *J. Chem. Phys.*, 88, (1988); Bartolotti L.J., Flurchick K., An introduction to density functional theory, *Rev. Computat. Chem.*, 7, pp. 187-216, (2007); Burke K., Perdew J.P., Wang Y., Derivation of a generalized gradient approximation: the PW91 density functional, *Electronic Density Functional Theory*, (1998); Grimme S., Neese F., Double hybrid density functional theory for excited states of molecules, *J. Chem. Phys.*, 127, (2007); Ghosh S., Verma P., Cramer C.J., Gagliardi L., Truhlar D.G., Combining Wave Function Methods with Density Functional Theory for Excited States, *Chem Rev*, (2018); Neese F., (2020); Handy N.C., Cohen A.J., Left-right correlation energy, *Mol. Phys.*, 99, pp. 403-412, (2001); Perdew J.P., Burke K., Ernzerhof M., Generalized gradient approximation made simple, *Phys. Rev. Lett.*, 77, pp. 3865-3868, (1996); Swart M., Ehlers A.W.,

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Lammertsma K., Performance of the OPBE exchange-correlation functional, *Mol. Phys.*, 102, pp. 2467-2474, (2004); Zhang Y., Wu A., Xu X., Yan Y., OPBE: a promising density functional for the calculation of nuclear shielding constants, *Chem. Phys. Lett.*, 421, pp. 383-388, (2006); Kohout M., (2011); Savin A., Nesper R., Wengert S., Fassler T.F., ELF: The electron localization function, *Angew. Chem. Int. Ed. Engl.*, 36, pp. 1808-1832, (1997); Zarate X., Pop L.C., Treto-Suarez M., Tapia J., Schott E., Structure and electronic properties of benzimidazole and cycloheptaimidazole gold N-heterocyclic carbenes, *Polyhedron*, 205, (2021); Becke A.D., Edgecombe K.E., A simple measure of electron localization in atomic and molecular systems, *J. Chem. Phys.*, 92, pp. 5397-5403, (1990); Jaque P., Toro-Labbe A., Characterization of copper clusters through the use of density functional theory reactivity descriptors, *J. Chem. Phys.*, 117, 7, pp. 3208-3220, (2002); Jan Moens P.G., Jaque P., De Proft F., The study of redox reactions on the basis of conceptual DFT principles: EEM and vertical quantities, *J. Phys. Chem. A*, 112, pp. 6023-6031, (2008); Szentpaly L., Electrophilicity Index, *Am. Chem. Soc.*, 121, pp. 1922-1924, (1999); Andrew J.D.W.S., Jupp R., Timothy C., The global electrophilicity index as a metric for Lewis acidity, *Dalton Trans.*, 47, pp. 7029-7035, (2018); Ximena Zarate E.S., Pop L.-C., Treto-Suarez M., Tapia J., Structure and electronic properties of benzimidazole and cycloheptaimidazole Gold N-heterocyclic Carbenes, *Polyhedron*, 205, (2021); Baerends E.J., Ziegler T., Autschbach J., Bashford D.B.A., Bickelhaupt F.M., Bo C., Boerrigter P.M., Cavallo L., Chong D., Et al., ADF2017, SCM, *Theoret. Chem.*, (2017); Mitoraj M., Michalak A., Ziegler T., In the nature of the agostic bond between metal centers and beta-hydrogen atoms in alkyl complexes. An analysis based on the extended transition state method and the natural orbitals for chemical valence scheme (ETS-NOCV), *Organometallics*, 28, (2009); Mitoraj M., Michalak A., Natural orbitals for chemical valence as descriptors of chemical bonding in transition metal complexes, *J. Mol. Model.*, 13, pp. 347-355, (2007); Mitoraj M.P., Bonding in ammonia borane: An analysis based on the natural orbitals for chemical valence and

---

the extended transition state method (ETS-NOCV), *J. Phys. Chem. A*, 115, pp. 14708-14716, (2011); Parr R.G., Pearson R.G., Absolute hardness: companion parameter to absolute electronegativity, *J. Am. Chem. Soc.*, 105, pp. 7512-7516, (1983); Pearson R.G., *Chemical hardness*, First, (1997); Reed J.L., Electronegativity: chemical hardness I, *Phys. Chem. A*, 101, pp. 7396-7400, (1997); David Pegu U.S., Deb J., Van Alsenoy C., Theoretical investigation of electronic, vibrational and nonlinear optical properties of 4-fluoro-4-hydroxybenzophenone, *Spectrosc. Lett.*, 50, pp. 232-243, (2017); Migahed M.A., Zaki E.G., Shaban M.M., Corrosion control in oil wells tubing steel during matrix acidizing operations, *Roy. Soc. Chem. Adv.*, 6, pp. 71384-71396, (2016); Aihara J., Correlation found between the HOMO-LUMO energy separation and the chemical reactivity at the most reactive site for isolated-pentagon isomers of fullerenes, *Phys. Chem. Chem. Phys.*, 2, pp. 3121-3125, (2000); Arnd Vogler H.K., Photochemistry induced by metal-to-ligand charge transfer excitation, *Coord. Chem. Rev.*, 208, pp. 321-329, (2000); Thomas Y.J.T., Allison C., Application of the condensed Fukui function to predict reactivity in core-shell transition metal nanoparticles, *Electrochim. Acta*, 101, pp. 334-340, (2013); Pucci G.G.N.A.R., Density functional theory, chemical reactivity, and the Fukui functions, *Found. Chem.*, 24, pp. 59-71, (2022); Luis Humberto Mendoza-Huizar, Chemical reactivity of quinmerac herbicide through the Fukui function, *Acta Chim. Slov.*, 61, pp. 694-702, (2014); Caramori G.F., Piccoli R.M., Segala M., Munoz-Castro A., Guajardo-Maturana R., Andrada D.M., Frenking G., Cyclic trinuclear copper(*i*), silver(*i*), and gold(*i*) complexes: a theoretical insight, *Dalton Trans.*, 44, pp. 377-385, (2015); Isaline Renard S.J.A., *Advances in Inorganic Chemistry and Radiochemistry*, (2020); Deborah M.D., Citrin E., Recent developments in radiotherapy, *Engl. J. Med.*, 377, pp. 1065-1075, (2017); Frederick Hawthorne A.M.M., Applications of radiolabeled boron clusters to the diagnosis and treatment of cancer, *Chem. Rev.*, 99, pp. 3421-3434, (1999); Fujii S., Expanding the chemical



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space of hydrophobic pharmacophores: the role of hydrophobic substructures in the development of novel transcription modulator, *Med. Chem. Commun.*, **7**, pp. 1082-1092, (2016); Wan L., Zhou Q., Wang X., Wood T.E., Wang L., Duchesne P.N., Guo J., Yan X., Xia M., Li Y.F., Jelle A.A., Ulmer U., Jia J., Li T., Sun W., Ozin G.A., Cu<sub>2</sub>O nanocubes with mixed oxidation-state facets for (photo)catalytic hydrogenation of carbon dioxide, *Nat. Catal.*, **2**, pp. 889-898, (2019); Taherpour A.A., Jamshidi M., Rezaei O., Belverdi A.R., Photoinduced electron transfer process on emission spectrum of N,N'-bis(salicylidene)-1,2-phenylenediamine as a Mg<sup>2+</sup> cation chemosensor: a first principle DFT and TDDFT study, *J. Mol. Struct.*, **1161**, pp. 339-344, (2018); Jager M., Freitag L., Gonzalez L., Using computational chemistry to design Ru photosensitizers with directional charge transfer, *Coord. Chem. Rev.*, **304-305**, pp. 146-165, (2015); Kaur B., Kaur N., Kumar S., Colorimetric metal ion sensors - a comprehensive review of the years 2011-2016, *Coord. Chem. Rev.*, **358**, pp. 13-69, (2018); Sinha N., Wenger O.S., Photoactive metal-to-ligand charge transfer excited states in 3d<sup>6</sup> complexes with Cr<sup>0</sup>, Mn<sup>I</sup>, Fe<sup>II</sup>, and Co<sup>III</sup>, *J. Am. Chem. Soc.*, **145**, pp. 4903-4920, (2023); Dean A.S., Olson L., Williksen E.P., An experimentally based model of the peroxidase-NADH biochemical oscillator: an enzyme-mediated chemical switch, *J. Am. Chem. Soc.*, **117**, pp. 2-15, (1995)

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